# An Enantioselective Total Synthesis of (+)-Atisirene by Intramolecular Double Michael Reaction 

Masataka Ihara, Masahiro Toyota, and Keiichiro Fukumoto*<br>Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan<br>Tetsuji Kametani<br>Institute of Medicinal Chemistry, Hoshi University, Ebara 2-4-41, Shinagawa-ku, Tokyo 142, Japan


#### Abstract

1,6-Conjugate addition of 2-methoxybenzylmagnesium bromide to 4-methylene-3,5,5-trimethylcyclo-hex-2-enone (7), followed by hydrocyanation, gave ( $\pm$ )-(1RS,2SR)-2-[2-(2-methoxyphenyl)ethyl]-1,3,3-trimethyl-5-oxocyclohexanecarbonitrile (9), which was converted into the ( $E$ )- $\alpha, \beta$-unsaturated enone ester (22). Intramolecular double Michael reaction of compound (22) produced the A/B-cis tetracyclic compound (23) in $17 \%$ yield. On the other hand, Wieland-Miescher ketone ( + )-(27) was selectively transformed into the ( $E$ )- $\alpha, \beta$-unsaturated enone ester (46), whose intramolecular double Michael reaction furnished the A/B-trans compound (48) in $92 \%$ yield. ( + ) -15 -Norisoatisirene (52), which had previously been transformed into ( + )-atisirene (53), was synthesized from keto ester (48).


(-)-Atisirene (1a) and (-)-isoatisirene (1b), first isolated from Erythroxylon monogynum, are structurally related to diterpene alkaloids of the atisine group. ${ }^{1}$ The first total synthesis of the ( $\pm$ )-forms of both compounds had been accomplished by Ireland and his co-workers ${ }^{2}$ before the discovery of the natural products, and $(+)$-atisirene (53) was synthesized starting from abietic acid. ${ }^{3}$ Partial syntheses of $(-)$-atisirene (1a) and ( - )isoatisirene (1b) were also performed starting from ( - )-kaurene ${ }^{4}$ and isosteviol. ${ }^{5}$ However, a challenging problem for the stereocontrolled construction of the bicyclo[2.2.2]octane part in the skeleton remains to be solved. In our previous paper, we disclosed a highly stereoselective synthesis of tricyclo[6.2.2.0 ${ }^{1.6}$ ]dodecanes by the novel intramolecular double Michael reaction, in which the desired stereoisomer (4) was formed as the sole product via the lithium-chelated intermediate (3). ${ }^{6}$ Applying this strategy, we have planned a new route to atisirene (1a) via the tetracyclic keto ester (5) which could be obtained by the annelation of the ( $E$ )- $\alpha, \beta$-unsaturated enone ester (6) (Scheme 1). Here we report a formal enantioselective total synthesis of $(+)$-atisirene (53). ${ }^{7}$

(1a) 16.17 -unsaturated
(1b) 15.16 - unsaturated

(1)


Scheme 1.

First we examined the preparation of the substrate (6) by means of an intramolecular double Michael reaction from the dienone (7). According to Davis' procedure, ${ }^{8}$ the dienone (7) was treated with 2 -methoxybenzylmagnesium bromide in the presence of copper( $(\mathrm{I})$ bromide and the resulting $1,6-$ conjugate adduct was isomerised with acid to give the enone (8) in $65 \%$ yield. Introduction of carbon units at the C-3position of compound (8) was a difficult problem mainly due to the steric effect of the dimethyl groups at C-5. After numerous attempts, a one-carbon unit could be introduced by hydrocyanation. ${ }^{9}$ Under thermodynamically controlled conditions using potassium cyanide-ammonium chloride in dimethylformamide (DMF), ${ }^{10}$ potassium cyanide-acetone cyanohydrin-18-crown-6, ${ }^{11}$ and diethylaluminium cyanide, ${ }^{12}$ only one isomer (9) was obtained, in $75--78 \%$ yield. On the other hand, kinetically controlled conditions with hydrogen cyanide and triethylaluminium ${ }^{13}$ gave the above product (9) in $71 \%$ yield along with the stereoisomer (10) in $4.5 \%$ yield. It was expected that compound (9) would be a thermodynamically stable isomer since a cyano group is less bulky than a methyl group. The structure of compounds (9) and (10) was determined by the following reactions. Reduction of isomer (9) with di-isobutylaluminium hydride (DIBAL) followed by acidic treatment afforded in $65 \%$ yield the cyclic hemiacetal (11), m.p. 88-89 ${ }^{\circ} \mathrm{C}$, while isomer (10) was converted into the formyl alcohol (13) in $66 \%$ yield under the same reaction conditions. Oxidation of the cyclic hemiacetal (11) with silver carbonate on Celite ${ }^{14}$ produced the lactone (12) in $80 \%$ yield (Scheme 2). Furthermore one of the methyl groups of the aldehyde (16), which was synthesized from compound (9) by dithioacetalisation in $92 \%$ yield, followed by reduction with DIBAL in $85 \%$ yield and desulphurisation with Raney nickel in $60 \%$ yield,
resonated at highfield, $\delta_{H} 0.73$, in the ${ }^{1} \mathrm{H}$ n.m.r. spectrum; the value indicated that one of the methyl groups at C-5 was shielded by the axially oriented formyl group. It was therefore revealed that the main product obtained by the hydrocyanation was the wrong precursor for the synthesis of atisirene. However, we carried on with this isomer in order to examine the validity of the intramolecular double Michael reaction for the assembly of the highly crowded compound.

Emmons reaction ${ }^{15}$ of the aldehyde (16) produced the $(E)$ $\alpha, \beta$-unsaturated ester (17) in $95 \%$ yield as a single product. Reduction of ester (17) with DIBAL in diethyl ether gave, in $94 \%$ yield, the allylic alcohol (18), which was treated with sodium hydride to protect the hydroxy group from hydrogenolysis, and then subjected to Birch reduction carried out using lithium in the presence of $t$-butyl alcohol in liquid ammonia. After treatment of the crude product with hydro-

(7)
(8)
(9)
(9)

(10)

(13)
Scheme 2.



Scheme 3.




Scheme 4.
chloric acid, the enone (19) was obtained in $50 \%$ yield based on consumed aryl ether (18). Oxidation with pyridinium chlorochromate (PCC) in the presence of Florisil, followed by further oxidation of the resulting aldehyde (20), obtained in $83 \%$ yield, with sodium chlorite in the presence of sulphamic acid in aqueous t-butyl alcohol, ${ }^{16}$ and esterification of the acid (21) with methanolic sulphuric acid furnished the $\alpha, \beta$-unsaturated enone ester (22) in $72 \%$ yield. The intramolecular double Michael reaction of compound (22) was carried out with lithium hexamethyldisilazide (LiHMDS) in a mixture of $n$-hexane and diethyl ether ( $8: 1, \mathrm{v} / \mathrm{v}$ ) for 2 h at $-78^{\circ} \mathrm{C}$ and then for 4 h at room temperature. After chromatographic separation, one isomer (23) of the possible tetracyclic compound was isolated in
$17 \%$ yield and no formation of other isomers was observed on careful inspection. The stereostructure (23) of the product was tentatively assigned on the basis of mechanistic considerations according to previous results. ${ }^{6}$ It is noteworthy that the cyclised product was obtained in spite of a crowded intermediate.

The ester moiety of compound (23) was removed by the following three steps; reduction of (23) with DIBAL in diethyl ether followed by oxidation of the resulting diol with pyridinium dichromate (PDC) in DMF to the keto aldehyde (24), m.p. $121-123{ }^{\circ} \mathrm{C}$, obtained in $55 \%$ yield from (23), and then decarbonylation with tris(triphenylphosphine)rhodium(1) chloride (TTPRCl) ${ }^{17}$ to give ketone (25) in $28 \%$ yield. Methylation of the ketone (25), m.p. $80-82^{\circ} \mathrm{C}$, with methyl
iodide in the presence of lithium di-isopropylamide (LDA) at 78 to $0^{\circ} \mathrm{C}$ afforded the tetramethylated compound (26) as a single isomer (Scheme 3). It was assumed on the basis of CPK molecular model considerations that the kinetically controlled alkylation would occur from the less hindered $\alpha$-side. The signals due to one of the quaternary methyl groups of compounds (25) and (26) were observed at lower field, $\delta_{\mathrm{H}} 1.14$, which would support the proposed structures. ${ }^{18}$

Now our attention focussed on the stereoselective construction of the a/b-trans isomer. We considered that the proper substrate (46) of the intramolecular double Michael reaction could be prepared from a trans-octahydronaphthalen-1(2H)one derivative. Thus the easily available Wieland-Miescher ketone $(+)-(27){ }^{19}$ was selectively converted into the ketone (33) by a modification of Sondheimer's method. ${ }^{20}$ Bromination of ketone (33) with pyridinium bromide perbromide (PBB) in acetic acid gave the $\alpha$-bromo ketone (34), m.p. 126-128 ${ }^{\circ} \mathrm{C}$, in $97 \%$ yield. In the n.m.r. spectrum of compound (34), $2-\mathrm{H}$ was observed as double doublet with $J 5.7$ and 11.4 Hz at $\delta_{\mathrm{H}} 4.94$; this indicated the presence of the equatorially oriented bromine atom. Treatment of compound (34) with sodium hydroxide in aqueous $\mathrm{DMF}^{21}$ at room temperature afforded, in $97 \%$ yield, the $\alpha$-hydroxy ketone (35), which was subjected to oxidative cleavage with lead tetra-acetate (LTA) in methanol followed by reduction of the resulting aldehyde, obtained in $90 \%$ yield, with sodium borohydride to furnish the alcohol (36) in $97 \%$ yield. The primary alcohol function of compound (36) was, on the action of 2-nitrophenyl selenocyanate and tributylphosphine, ${ }^{22}$ converted into the selenide (37) in $98 \%$ yield, oxidation of which with $30 \%$ hydrogen peroxide gave the olefin (38) in $91 \%$ yield. The ester group of compound (38) was reduced with lithium aluminium hydride to give the alcohol (39) in $95 \%$ yield. After oxidation of compound (39) with PCC in the presence of Florisil, the aldehyde, obtained in $88 \%$ yield, was protected as the ethylene acetal. The olefinic acetal (40), obtained in $98 \%$ yield, was oxidatively cleaved with osmium tetraoxide and sodium periodate ${ }^{23}$ to provide the unstable aldehyde (41) in $78 \%$ yield. Aldol condensation of compound (41) with cyclohexanone in the presence of LDA, carried out at between $-78^{\circ} \mathrm{C}$ and room temperature in tetrahydrofuran (THF), gave the ( $E$ )-enone (42), m.p. $52-53^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{17}-10.0^{\circ}\left(\mathrm{CHCl}_{3}\right)$, in $62 \%$ yield. Catalytic hydrogenation of enone (42) in the presence of $10 \%$ palladium-charcoal led quantitatively to the ketone (43), which was silylated under kinetically controlled conditions. Oxidation of the corresponding silyl enol ether with palladium(II) acetate in the presence of $p$-benzoquinone ${ }^{24}$ furnished the enone (44) in $85 \%$ yield. Deprotection of compound (44) with $10 \%$ perchloric acid in THF, followed by Emmons reaction ${ }^{15}$ of the aldehyde (45), obtained in $83 \%$ yield, using trimethyl phosphonoacetate [methyl (dimethoxyphosphoryl)acetate] in the presence of sodium hydride provided the ( $E$ )-x, $\beta$-unsaturated enone ester (46) in $47 \%$ yield. The intramolecular double Michael reaction was conducted using (LiHMDS) in a mixture of $n$-hexane and diethyl ether during 2.5 h at between $-78{ }^{\circ} \mathrm{C}$ and room temperature. The desired tetracyclic compound (48) was obtained as crystals, 145$148^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{25}-9.3^{\circ}$ (c 0.29 in $\mathrm{CHCl}_{3}$ ); in $92 \%$ yield. The highly stereocontrolled formation of compound (48) in such excellent yield can be ascribed to the lithium-chelated intermediate (47) having no strong non-bonding interaction. The structure of compound (48) was determined by its conversion into the known compounds (50)-(52). In the same manner as previously, reduction of ester (48) with DIBAL followed by oxidation of the corresponding diol with PDC and decarbonylation of the aldehyde (49), obtained in $59 \%$ from ester (48), using TTPRCl ${ }^{17}$ gave, in $28 \%$ yield, the $(+)$-ketone (50), m.p. $147-149^{\circ} \mathrm{C}$ (lit., ${ }^{3} 146-148^{\circ} \mathrm{C}$ ); $[\alpha]_{\mathrm{D}}^{25}+19.5^{\circ}(c$ 0.021 in MeOH$)\left\{\right.$ lit., ${ }^{3}[\alpha]_{\mathrm{D}}+19.6^{\circ}(c 0.05$ in MeOH$\left.)\right\}$, whose
spectral data were similar to reported ones. ${ }^{3}$ Methylation of compound (50) with methyl iodide in the presence of LDA provided, in $38 \%$ yield, atisiran- 15 -one (51), whose n.m.r. data were consistent with those of the authentic compound. ${ }^{25,26}$ The transformation of the ketone (50) into the olefin (52) was accomplished in the following three steps in $70 \%$ overall yield; bromination of compound ( $\mathbf{5 0}$ ) with bromine and LDA, sodium borohydride reduction of the resulting bromo ketone, and reduction of the resulting bromohydrin with zinc in ethanol. ${ }^{27}$ The n.m.r. spectral data of (+)-15-norisoatisirene (52), m.p. $83-84^{\circ} \mathrm{C}$ (lit., ${ }^{3} \quad 82-83^{\circ} \mathrm{C}$ ), $\quad[\alpha]_{\mathrm{D}}^{24}+16.4^{\circ}$ (c 0.134 in MeOH ), agreed well with those reported. ${ }^{3}$ Since compound (52) had been converted into ( + )-atisirene (53) in three steps, ${ }^{3}$ its formal enantioselective synthesis has been achieved (Scheme 4).

## Experimental

General Methods.-M.p.s are uncorrected. I.r. spectra were recorded on a Hitachi 260-10 spectrophotometer for solutions in $\mathrm{CHCl}_{3}$. N.m.r. spectra were measured on a JEOL JNM-PMX-60 or a JEOL-PS-100 spectrometer for solutions in $\mathrm{CDCl}_{3}$ unless otherwise stated. Chemical shifts are reported as $\delta_{\mathrm{H}}$ values relative to internal $\mathrm{SiMe}_{4}$. Ordinary mass spectra were taken on a Hitachi M-52G machine, and accurate mass spectra with a JEOL-JMS-01SG-2 spectrometer. All new compounds described in the Experimental section were homogeneous on t.l.c. Magnesium sulphate was used to dry extracts. High-pressure liquid chromatography was carried out using a Hitachi 635 instrument monitored by u.v. absorption and refractive-index measurements. Optical rotations were measured on a JASCO-PIP-SL polarimeter for solutions in $\mathrm{CHCl}_{3}$ unless otherwise stated.

4-[2-(2-Methoxyphenyl)ethyl]-3,5,5-trimethylcyclohex-2enone (8).-A solution of the Grignard reagent prepared from 2-methoxybenzyl bromide ( $26.8 \mathrm{~g}, 0.13 \mathrm{~mol}$ ) and magnesium $(4.2 \mathrm{~g}, 0.17 \mathrm{~mol})$ in dry THF ( 80 ml ) was slowly added to a stirred mixture of the dienone (7) $(10 \mathrm{~g}, 66 \mathrm{mmol})$ and copper(I) bromide ( $1 \mathrm{~g}, 3.5 \mathrm{mmol}$ ) in dry diethyl ether $(80 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. The mixture was allowed to warm to room temperature, stirred for 0.5 h at the same temperature under nitrogen, and then poured into saturated aqueous ammonium chloride ( 200 ml ). The resulting mixture was extracted with diethyl ether. The extract was concentrated to give a residue, which was taken up into THF ( 50 ml ). To the mixture was added $10 \%$ hydrochloric acid ( 50 ml ) and the resulting mixture was refluxed for 2 h under nitrogen. Extraction of the mixture with diethyl ether, followed by drying and evaporation of the extract, gave a residue, which was purified by chromatography on silica gel with $n$-hexaneacetone ( $50: 1, \mathrm{v} / \mathrm{v}$ ) as eluant to afford the enone ( 8 ) ( 11.8 g , $65 \%$ ) as an oil (Found: C, 79.3; H, 8.75. $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{O}_{2}$ requires C, $79.35 ; \mathrm{H}, 8.9 \%$; $v_{\text {max. }} 1650 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 1.00$ and 1.10 (each 3 H , each s, $2 \times \mathrm{Me}$ ), $1.90(3 \mathrm{H}, \mathrm{s}, 3-\mathrm{Me}), 3.80(3$ $\mathrm{H}, \mathrm{s}, \mathrm{OMe}), 5.63(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 2-\mathrm{H})$, and $6.57-7.23(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$; $m / z 272\left(M^{+}\right)$.
( $\pm$ )-(1RS,2SR)- and (1RR,2RR)-2-[2-(2-Methoxyphenyl)-ethyl]-1,3,3-trimethyl-5-oxocyclohexanecarbonitrile (9) and (10).-Method A. To a stirred solution of triethylaluminium $(255 \mathrm{mg}, 2.24 \mathrm{mmol})$ in dry THF ( 4 ml ) at $0^{\circ} \mathrm{C}$ was added a dry THF solution ( 2 ml ) of hydrogen cyanide $(40 \mathrm{mg}, 1.48$ $\mathrm{mmol})$ and the enone ( 8 ) ( $200 \mathrm{mg}, 0.74 \mathrm{mmol}$ ). The mixture was allowed to warm to room temperature, stirred for 22 h under argon, and then poured into a vigorously stirred mixture of 2 m -sodium hydroxide and ice. The resulting solution was extracted with diethyl ether and the extract was washed successively with water, 2 m -hydrochloric acid, and water, dried, and evaporated to afford a crude product, which was subjected to column chromatography on silica gel. Elution
with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) gave the nitrile (10) ( $10 \mathrm{mg}, 4.5 \%$ ) as an oil, $v_{\text {max. }} 2225(\mathrm{CN})$ and $1720 \mathrm{~cm}^{-1}$ $(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.90,1.16$, and 1.27 (each 3 H , each $\mathrm{s}, 3 \times \mathrm{Me}$ ), $1.40-3.20\left(9 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{CH}_{2}\right.$ and CH ), $3.83(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, and 6.66-7.23 (4 H, m, ArH); m/z $299\left(M^{+}\right)$(Found: $M^{+}$, 299.1885. $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{NO}_{2}$ requires $M$, 299.1886).

Further elution gave the isomer (9) ( $157 \mathrm{mg}, 71 \%$ ) as an oil, $\nu_{\text {max. }} 2225(\mathrm{CN})$ and $1720 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 1.04,1.07$, and 1.54 (each 3 H , each s, $3 \times \mathrm{Me}$ ), 1.42-2.85 $\left(9 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{CH}_{2}\right.$ and $\mathrm{CH}), 3.83(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, and $6.57-7.27(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}) ; m / z$ $299\left(M^{+}\right)$(Found: $M^{+}$, 299.1893).

Method B. To a mixture of the enone (8) ( $1 \mathrm{~g}, 3.7 \mathrm{mmol}$ ), DMF ( 40 ml ), and water ( 6.3 ml ) were added potassium cyanide ( $480 \mathrm{mg}, 7.4 \mathrm{mmol}$ ) and ammonium chloride ( $300 \mathrm{mg}, 5.6$ mmol ). The resulting mixture was stirred and heated at $100^{\circ} \mathrm{C}$ for 72 h . After evaporation of the solvent, the resulting residue was taken up into diethyl ether, and the extract was washed with saturated aqueous ammonium chloride and dried. Evaporation of the solvent gave a crude oil, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $10: 3 \mathrm{v} / \mathrm{v}$ ) afforded the starting material (8) ( 136 mg ) and the nitrile (9) [ $722 \mathrm{mg}, 76 \%$ based on consumed (8)], whose spectral data and t.l.c. behaviour were identical with those of the sample prepared by method A .

Method C. To a stirred solution of the enone (8) $(5 \mathrm{~g}, 0.018$ mol ) in acetonitrile ( 50 ml ) were added potassium cyanide ( 1.2 $\mathrm{g}, 0.018 \mathrm{~mol}$ ), acetone cyanohydrin ( $1.57 \mathrm{~g}, 0.018 \mathrm{~mol}$ ), and 18 -crown-6 ( $4.8 \mathrm{~g}, 0.018 \mathrm{~mol}$ ) at ambient temperature. The mixture was refluxed for 120 h . Removal of the solvent under reduced pressure afforded a residue, which was purified by silica gel column chromatography with n-hexane-ethyl acetate ( $10: 3, \mathrm{v} / \mathrm{v}$ ) as eluant to afford the starting material (8) (790 mg ) and the nitrile (9) [ $3.5 \mathrm{~g}, 75.6 \%$ based on consumed (8)], whose spectral data and t.l.c. behaviour were identical with those of the sample prepared by method A.

Method D. To a mixture of the enone (8) $(120 \mathrm{mg}, 0.44 \mathrm{mmol})$ in dry benzene ( 5 ml ) and dry toluene ( 3 ml ) at $0^{\circ} \mathrm{C}$ under argon was added dropwise diethylaluminium cyanide $(294 \mathrm{mg}, 2.6$ mmol ). After being stirred for 0.5 h , the resulting mixture was poured into $4 \%$ aqueous sodium hydroxide $(50 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. The mixture was extracted with chloroform. The extract was dried and then evaporated to afford a residue, which was purified by chromatography on silica gel, with n-hexane-diethyl ether $(10: 3), \mathrm{v} / \mathrm{v}$ ) as eluant, to afford the nitrile (9) ( $103 \mathrm{mg}, 78 \%$ ), whose spectral data and t.l.c. behaviour were identical with those of the sample prepared by method A.
( $\pm$ )-(1RS,2RS,5SR)-5-Hydroxy-2-[2-(2-methoxyphenyl)-ethyl]-1,3,3-trimethylcyclohexanecarbaldehyde (13).-To a stirred solution of the nitrile ( 10 ) ( $3 \mathrm{mg}, 0.01 \mathrm{mmol}$ ) in dry 1,2-dimethoxyethane (DME) ( 2 ml ) at $-78^{\circ} \mathrm{C}$ under argon was added dropwise DIBAL $(4.3 \mathrm{mg}, 0.03 \mathrm{mmol})$. The mixture was allowed to warm to room temperature during 0.5 h and was then refluxed for 1 h . To the mixture was added slowly $10 \%$ hydrochloric acid ( 3 ml ), and the resulting mixture was heated under reflux for 1 h . After cooling to room temperature, the mixture was extracted with diethyl ether. The extract was washed with saturated aqueous sodium chloride, dried, and evaporated to afford a residue, which was subjected to column chromatography on silica gel. Elution with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) gave the hydroxy aldehyde ( 13 ) ( $2 \mathrm{mg}, 66 \%$ ) as an oil, $v_{\text {max }} 3600(\mathrm{OH})$ and $1723 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.87,1.03$, and 1.07 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $1.20-2.75(9 \mathrm{H}$, m, $4 \times \mathrm{CH}_{2}$ and CH$), 3.80(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.98-4.20(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H})$, $6.69-7.37(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, and $9.40(1 \mathrm{H}, \mathrm{s}, \mathrm{CHO}) ; m / z 304\left(\mathrm{M}^{+}\right)$.

## (土)-(1RS,2SR,5RS,7SR)-2-[2-(2-Methoxyphenyl)ethyl]-

 1,3,3-trimethyl-6-oxabicyclo[3.2.1]octan-7-ol (11).-Tostirred solution of the nitrile (9) ( $26 \mathrm{mg}, 0.087 \mathrm{mmol}$ ) in dry DME ( 3 ml ) at $-78^{\circ} \mathrm{C}$ under argon was added dropwise DIBAL ( $37 \mathrm{mg}, 0.26 \mathrm{mmol}$ ). The mixture was allowed to warm to room temperature during 0.5 h , and was then refluxed for 1 h . To the mixture was added slowly $10 \%$ hydrochloric acid ( 3 ml ), and the resulting mixture was heated under reflux for 1 h . After cooling to room temperature, the mixture was extracted with diethyl ether. The extract was washed with saturated aqueous sodium chloride, dried, and evaporated to afford a residue, which was subjected to column chromatography on silica gel. Elution with $n$-hexane-ethyl acetate ( $5: 1, \mathrm{v} / \mathrm{v}$ ) gave the cyclic hemiacetal ( 11 ) ( $17 \mathrm{mg}, 65 \%$ ) as needles, m.p. $88-89^{\circ} \mathrm{C} ; \delta_{\mathrm{H}} 0.91$, 1.03 , and 1.23 (each 3 H , each s, $3 \times \mathrm{Me}$ ), 3.81 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$ ), $4.43(1 \mathrm{H}, \mathrm{br} \mathrm{t}, J 5.7 \mathrm{~Hz}, 5-\mathrm{H}), 5.10(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 7-\mathrm{H})$, and $6.71-$ $7.36\left(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}\right.$ ); m/z $304\left(M^{+}\right)$(Found: $M^{+}$, 304.2036. $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{O}_{3}$ requires $M, 304.1995$ ).
( $\pm$ )-(1RS,2SR,5RS)-2-[2-(2-Methoxyphenyl)ethyl]-1,3,3-trimethyl-6-oxabicyclo[3.2.1]octan-7-one (12).-To a solution of compound ( 11 ) ( $5 \mathrm{mg}, 0.016 \mathrm{mmol}$ ) in dry benzene was added silver carbonate on Celite ( $80 \mathrm{mg}, 0.29 \mathrm{mmol}$ ) and the mixture was refluxed for 2 h . After cooling to room temperature, the mixture was filtered through Celite. Evaporation of the filtrate gave a residue, which was chromatographed on silica gel. Elution with n -hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) gave the lactone (12) $(4 \mathrm{mg}, 80 \%)$ as an oil, $v_{\text {max. }} 1760 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.93,0.99$, and 1.21 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $1.20-2.95(9 \mathrm{H}$, m, $4 \times \mathrm{CH}_{2}$ and CH$), 3.81(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 5.66(1 \mathrm{H}, \mathrm{brt}, J 5.7 \mathrm{~Hz}$, $5-\mathrm{H})$, and $6.70-7.31(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}) ; \mathrm{m} / \mathrm{z} 302\left(\mathrm{M}^{+}\right)$(Found: $M^{+}$, 302.1882. $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{O}_{3}$ requires $M, 302.1882$ ).
( $\pm$ )-(1RS,2SR)-5,5-Ethylenedithio-2-[2-(2-methoxyphenyl)-ethyl]-1,3,3-trimethylcyclohexanecarbonitrile (14).-To a stirred solution of the ketone (9) ( $1.13 \mathrm{~g}, 3.77 \mathrm{mmol}$ ) and ethane-1,2-dithiol ( $710 \mathrm{mg}, 7.55 \mathrm{mmol}$ ) in dry methylene dichloride ( 15 ml ) was added dropwise boron trifluoridediethyl ether ( $43 \mathrm{mg}, 0.377 \mathrm{mmol}$ ) at room temperature. After being stirred for 5 h at room temperature, the mixture was treated with saturated sodium chloride ( 10 ml ), and the resulting mixture was extracted several times with methylene dichloride. The combined extract was dried and evaporated to afford a powder, which was recrystallised from benzenen -hexane ( $2: 5, \mathrm{v} / \mathrm{v}$ ) to give the thioacetal (14) ( $1.30 \mathrm{~g}, 92 \%$ ) as needles, m.p. 163-164 ${ }^{\circ} \mathrm{C}$ (Found: C, 66.8; H, 7.6; N, 3.6; S, 16.9. $\mathrm{C}_{12} \mathrm{H}_{29} \mathrm{NOS}_{2}$ requires $\mathrm{C}, 67.15 ; \mathrm{H}, 7.8 ; \mathrm{N}, 3.75$; S , $17.05 \%) ; v_{\text {max. }} 2220 \mathrm{~cm}^{-1}(\mathrm{C} \equiv \mathrm{N}) ; \delta_{\mathrm{H}} 0.94,1.28$, and 1.45 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $1.50-2.90\left(9 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{CH}_{2}\right.$ and CH$)$, 3.17-3.57 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{SCH}_{2} \mathrm{CH}_{2} \mathrm{~S}$ ), $3.80(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, and 6.63-7.34 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ); m/z $375\left(M^{+}\right)$.
( $\pm$ )-(1RS,2SR)-5,5-Ethylenedithio-2-[2-(2-methoxyphenyl)-ethyl]-1,3,3-trimethylcyclohexanecarbaldehyde (15).-To a stirred solution of the nitrile (14) ( $3.0 \mathrm{~g}, 8 \mathrm{mmol}$ ) in dry DME $(30 \mathrm{ml})$ under argon was added dropwise DIBAL ( $2.27 \mathrm{~g}, 16$ mmol ) at ambient temperature, and the mixture was refluxed for 2 h . To the resulting mixture was added slowly $10 \%$ hydrochloric acid ( 50 ml ) and the mixture was heated under reflux for 2 h . After cooling to room temperature, the mixture was extracted with diethyl ether. The extract was washed with saturated aqueous sodium chloride, dried, and evaporated to afford a residue, which was purified by silica gel column chromatography. Elution with n-hexane-diethyl ether ( $2: 1, \mathrm{v} / \mathrm{v}$ ) afforded the aldehyde (15) ( $2.57 \mathrm{~g}, 85 \%$ ) as an oil, $v_{\text {max. }} 1710$ $\mathrm{cm}^{-1}(\mathrm{C}=\mathrm{O})$ ) $\delta_{\mathrm{H}} 1.05(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Me}), 1.08(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$, 3.18-3.35 (4 H, m, SCH $\left.\mathrm{CH}_{2} \mathrm{~S}\right), 3.78(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 6.63-7.30$ ( $4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ), and 9.77 ( $1 \mathrm{H}, \mathrm{s}, \mathrm{CHO}$ ); m/z $378\left(\mathrm{M}^{+}\right.$) (Found: $M^{+}, 378.1688 . \mathrm{C}_{21} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{~S}_{2}$ requires $M, 378.1706$ )
( $\pm$ )-(1 RS,2SR)-2-[2-(2-Methoxyphenyl)ethyl]-1,3,3-trimethylcyclohexanecarbaldehyde (16).-A mixture of the thioacetal (15) ( $50 \mathrm{mg}, 0.13 \mathrm{mmol}$ ) and $\mathrm{W}_{1}$-Raney nickel ( 3 g ) in acetone ( 7 ml ) was stirred for 0.5 h at room temperature and then filtered through Celite. After evaporation of the filtrate, the residue was purified by column chromatography on silica gel. Elution with n-hexane-ethyl acetate ( $10: 1$, v/v) yielded the aldehyde (16) ( $23 \mathrm{mg}, 60 \%$ ) as an oil (Found: C, $78.8 ; \mathrm{H}$, 10.0. $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{O}_{2}$ requires $\mathrm{C}, 79.1 ; \mathrm{H}, 9.8 \%$ ); $v_{\text {max. }} 1710 \mathrm{~cm}^{-1}$ $(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.73,0.99$, and 1.02 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $0.95-2.85\left(11 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right.$ and CH$), 3.80(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, $6.58-7.23(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, and $9.67(1 \mathrm{H}, \mathrm{s}, \mathrm{CHO}) ; m / z 288$ $\left(M^{+}\right)$.

Ethyl ( $\pm$ )-( $1^{\prime}$ RS, $2^{\prime}$ RS)-3-\{-2'-[2-(2-Methoxyphenyl)ethyl]$1^{\prime}, 3^{\prime}, 3^{\prime}$-trimethylcyclohexyl $\}$ prop-2(E)-enoate (17)--To a stirred suspension of $60 \%$ sodium hydride ( $28 \mathrm{mg}, 0.69 \mathrm{mmol}$ ) in dry DME ( 3 ml ) was added dropwise triethyl phosphonoacetate [ethyl (diethoxyphosphoryl)acetate] ( $156 \mathrm{mg}, 0.69 \mathrm{mmol}$ ) at room temperature. After being stirred for 0.5 h , the mixture was treated dropwise a solution of the aldehyde (16) ( $100 \mathrm{mg}, 0.35$ mmol ) in dry DME ( 2 ml ). The resulting mixture was heated for 4 h under reflux. After addition of large amount of water at $0^{\circ} \mathrm{C}$, the mixture was extracted with diethyl ether. The extract was dried and evaporated to give a residue, which was chromatographed on silica gel. Elution with n-hexane-diethyl ether ( $20: 1, \mathrm{v} / \mathrm{v}$ ) afforded the $\alpha, \beta$-unsaturated ester ( 17 ) ( $116 \mathrm{mg}, 95 \%$ ) as an oil (Found: C, 77.2; H, 9.3. $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{O}_{3}$ requires $\mathrm{C}, 77.05$; $\mathrm{H}, 9.55 \%$ ); $v_{\text {max. }} 1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.77,0.93$, and 1.05 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $1.23\left(3 \mathrm{H}, \mathrm{t}, J 7 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Me}\right)$, $3.82(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 4.12\left(2 \mathrm{H}, \mathrm{q}, J 7 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Me}\right), 5.68(1 \mathrm{H}, \mathrm{d}$, $\left.J 16 \mathrm{~Hz},=\mathrm{CHCO}_{2} \mathrm{Et}\right)$, and $6.63-7.35(5 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ and olefinic H); $m / z 358\left(M^{+}\right)$.
( $\pm$ )-( $\left.1^{\prime} \mathrm{RS}, 2^{\prime} \mathrm{RS}\right)-3-\left\{2^{\prime}-[2-(2-M e t h o x y p h e n y l) e t h y l]-1^{\prime}, 3^{\prime}, 3^{\prime}-\right.$ trimethylcyclohexyl\}prop-2(E)-enol (18).-To a stirred solution of the ester (17) ( $3.0 \mathrm{~g}, 8.38 \mathrm{mmol}$ ) in dry diethyl ether ( 30 ml ) at $-78{ }^{\circ} \mathrm{C}$ was added dropwise DIBAL ( $2.38 \mathrm{~g}, 16.76 \mathrm{mmol}$ ). After 1 h , the mixture was allowed to warm to room temperature and was then stirred for 1 h at the same temperature. To the stirred solution was added water ( 2.4 ml ) at $0^{\circ} \mathrm{C}$, and the mixture was stirred for 0.5 h . The resulting mixture was filtered through Celite and the filtrate was evaporated to give a residue, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) gave the allylic alcohol (18) $(2.5 \mathrm{~g}, 94 \%)$ as an oil (Found: C, 80.1; H, 10.4. $\mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{2}$ requires C, $79.7 ; \mathrm{H}, 10.2 \%$ ); $v_{\text {max. }} 3600 \mathrm{~cm}^{-1}(\mathrm{OH})$; $\delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.77,0.93$, and 1.00 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $0.90-$ $2.85\left(11 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right.$ and CH$), 3.80(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.97(2 \mathrm{H}$, br d, $\left.J 4 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{OH}\right), 5.30-5.73\left(1 \mathrm{H}, \mathrm{m},=\mathrm{CHCH}_{2}\right), 5.87(1 \mathrm{H}$, d, $\left.J 16 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CHCH}_{2}\right)$, and $6.62-7.30(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}) ; m / z 316$ $\left(M^{+}\right)$.
( $\pm$ )-(1"RS,2"RS)-6-\{2-[3-Hydroxyprop-1(E)-enyl]-
$2^{\prime \prime}, 6^{\prime \prime}, 6^{\prime \prime}$-trimethylcyclohexylethyl $\}$ cyclohex-2-enone (19).-To a solution of the alcohol (18) ( $43 \mathrm{mg}, 0.14 \mathrm{mmol}$ ) in dry THF ( 3 ml ) was added $60 \%$ sodium hydride ( $8 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) and the mixture was stirred for 30 min at room temperature. Liquid ammonia ( 6 ml ), t-butyl alcohol ( 3 ml ), and lithium ( $30 \mathrm{mg}, 4.3$ mmol ) were added to the above mixture, and the resulting mixture was stirred for 2 h under nitrogen. After evaporation of the solvents, followed by addition of saturated aqueous ammonium chloride, the mixture was extracted with diethyl ether. The extract was dried and evaporated to give an oil, which was dissolved in methylene dichloride ( 2.5 ml ). After addition of $10 \%$ hydrochloric acid $(2.5 \mathrm{ml})$, the mixture was stirred for 15 h under reflux. The mixture was extracted with methylene dichloride, and the extract was washed with
saturated aqueous sodium chloride, dried, and evaporated to give a residue, which was purified by silica gel chromatography. Elution with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) gave the starting material (18) ( 10 mg ) and the enone (19) [17 mg, $50 \%$ based on consumed (18)] as an oil, $v_{\text {max. }} 3600(\mathrm{OH})$ and 1670 $\mathrm{cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.77,0.86$, and 0.95 (each 3 H , each s, $3 \times \mathrm{Me}), 4.03\left(2 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, \mathrm{C} \mathrm{H}_{2} \mathrm{OH}\right), 5.30-5.86(2 \mathrm{H}, \mathrm{m}$, $=\mathrm{CHCH}_{2} \mathrm{OH}$ and $=\mathrm{CHCO}$ ), $5.87(1 \mathrm{H}$, br d, $J 16 \mathrm{~Hz}$, $\mathrm{CH}=\mathrm{CHCH}_{2}$ ), and 6.65-7.00 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CHCO}$ ); $m / z 304$ ( $M^{+}$) (Found: $M^{+}, 304.2403 . \mathrm{C}_{20} \mathrm{H}_{32} \mathrm{O}_{2}$ requires $M, 304.2405$ ).
( $\pm$ )-( $\left.1^{\prime} \mathrm{RS}, 2^{\prime} \mathrm{RS}\right)-(\mathrm{E})-3-\left\{1^{\prime}, 3^{\prime}, 3^{\prime}\right.$-Trimethyl-2'-[2-(2-oxocyclo-hex-3-enyl)ethyl]cyclohexyl\}acrylaldehyde (20).-To a solution of the alcohol (19) ( $134 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) in dry methylene dichloride ( 5 ml ) were added Florisil ( 190 mg ) and PCC ( $190 \mathrm{mg}, 0.88 \mathrm{mmol}$ ), and the mixture was stirred for 3 h at room temperature. Filtration, followed by evaporation of the filtrate, gave a residue, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $10: 2, \mathrm{v} / \mathrm{v}$ ) afforded the aldehyde (20) ( $110 \mathrm{mg}, 83 \%$ ) as an oil, $v_{\text {max. }} 1670 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O})$; $\delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.75,0.90$, and 1.07 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $0.95-$ $2.50\left(15 \mathrm{H}, \mathrm{m}, 7 \times \mathrm{CH}_{2}\right.$ and CH$), 5.73-6.23(2 \mathrm{H}, \mathrm{m},=\mathrm{CHCHO}$ and $=\mathrm{CHCO}$ ), $6.67-7.03(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CHCO}), 6.97(1 \mathrm{H}, \mathrm{d}, \mathrm{J}$ $16 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CHCHO}$ ), and 9.47 ( $1 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, \mathrm{CHO}$ ); m/z 302 ( $M^{+}$) (Found: $M^{+}$, 302.2247. $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{O}_{2}$ requires $M, 302.2278$ ).
( $\pm$ )-( $\left.1^{\prime} \mathrm{RS}, 2^{\prime} \mathrm{RS}\right)-3-\left\{1^{\prime}, 3^{\prime}, 3^{\prime}\right.$-Trimethyl-2'-[2-(2-oxocyclohex-3-enyl)ethyl]cyclohexyl\}prop-2(E)-enoic Acid (21).-To a stirred mixture of the aldehyde (20) ( $100 \mathrm{mg}, 0.3 \mathrm{mmol}), \mathrm{t}$-butyl alcohol $(0.5 \mathrm{ml})$, and water ( 4 ml ) were added sulphamic acid $(400 \mathrm{mg}, 4.1 \mathrm{mmol})$ and sodium chlorite $(400 \mathrm{mg}, 4.4 \mathrm{mmol})$, and the mixture was stirred for 1 h at room temperature. After addition of water ( 4 ml ), the mixture was extracted with diethyl ether. The extract was washed with water, dried, and evaporated to give a residue, which was subjected to silica gel column chromatography. Elution with n -hexane-diethyl ether ( $1: 2, \mathrm{v} / \mathrm{v}$ ) gave the carboxylic acid (21) (74 mg, 70\%) as a syrup, $v_{\text {max. }} 1690$ and $1670 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.73,0.90$, and 1.05 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $0.98-2.55\left(15 \mathrm{H}, \mathrm{m}, 7 \times \mathrm{CH}_{2}\right.$ and CH$), 5.70(1$ $\left.\mathrm{H}, \mathrm{d}, J 16 \mathrm{~Hz},=\mathrm{CHCO}_{2} \mathrm{H}\right), 5.93(1 \mathrm{H}, \mathrm{brd}, J 10 \mathrm{~Hz},=\mathrm{CHCO})$, 6.62-7.03 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}=$ ), $7.28(1 \mathrm{H}, \mathrm{d}, J 16 \mathrm{~Hz}$, $\left.\mathrm{CH}=\mathrm{CHCO}_{2} \mathrm{H}\right)$, and $10.90\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CO}_{2} \mathrm{H}\right) ; m / z 318\left(M^{+}\right)$ (Found: $M^{+}, 318.2193 . \mathrm{C}_{20} \mathrm{H}_{30} \mathrm{O}_{3}$ requires $M, 318.2165$ ).

Methyl ( $\pm$ )-( $\left.1^{\prime} \mathrm{RS}, 2^{\prime} \mathrm{RS}\right)-3-\left\{1^{\prime}, 3^{\prime}, 3^{\prime}\right.$-Trimethyl-2'-[2-(2-oxo-cyclohex-3-enyl)ethyl]cyclohexyl\}prop-2(E)-enoate To a solution of the acid (21) $(10 \mathrm{mg}, 0.03 \mathrm{mmol})$ in anhydrous methanol ( 1 ml ) at $0^{\circ} \mathrm{C}$ was slowly added conc. sulphuric acid $(0.05 \mathrm{ml})$, and the mixture was stirred for 1 h at room temperature. After addition of water ( 5 ml ), the mixture was extracted with diethyl ether. The extract was washed with saturated aqueous sodium hydrogen carbonate, dried, and evaporated to give a residue, which was purified by silica gel column chromatography. Elution with n-hexane-diethyl ether ( $10: 1, \mathrm{v} / \mathrm{v}$ ) afforded the ester ( 22 ) ( $7.5 \mathrm{mg}, 72 \%$ ) as an oil, $v_{\text {max. }}$. 1710 and $1670 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.74,0.89$, and 1.01 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $1.05-2.50\left(15 \mathrm{H}, \mathrm{m}, 7 \times \mathrm{CH}_{2}\right.$ and CH$), 3.73$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 5.77\left(1 \mathrm{H}, \mathrm{d}, J 16 \mathrm{~Hz},=\mathrm{CHCO}_{2} \mathrm{Me}\right), 5.97(1 \mathrm{H}, \mathrm{dt}$, $J 2$ and $\left.10 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}\right), 6.91\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}=\right)$, and 7.19 $\left(1 \mathrm{H}, \mathrm{d}, J 16 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CHCO}_{2} \mathrm{Me}\right.$ ); $m / z 332\left(M^{+}\right)$(Found: $M^{+}$, 332.2350. $\mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{3}$ requires $M, 332.2349$ ).
( $\pm$ )-(3RS,4SR,4aSR,4bRS,8aSR,10aSR)-1,2,3,4,4a,4b,5,6,7,-8,8a,9,10,10a-Tetradecahydro-4b,8,8-trimethyl-1-oxo-3,10a-ethanophenanthrene-4-carboxylate (23).-To a stirred solution of LiMHDS, prepared from 1,1,1,3,3,3-hexamethyldisilazane ( $19 \mathrm{mg}, 0.12 \mathrm{mmol}$ ) and butyl-lithium ( $7.5 \mathrm{mg}, 0.12 \mathrm{mmol}$ ), in dry n -hexane ( 4 ml ) at $-78^{\circ} \mathrm{C}$ under argon was added a
solution of the ester ( 22 ) ( $30 \mathrm{mg}, 0.09 \mathrm{mmol}$ ) in dry diethyl ether $(0.5 \mathrm{ml})$, and the mixture was stirred for 2 h at $-78^{\circ} \mathrm{C}$, and for 4 $h$ at ambient temperature. After the reaction mixture had been poured into saturated aqueous ammonium chloride ( 3 ml ), the mixture was extracted with diethyl ether. The extract was dried and evaporated to give a residue, which was chromatographed on silica gel. Elution with $n$-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) afforded the tetracyclic compound (23) ( $5 \mathrm{mg}, 17 \%$ ) as needles, m.p. $123-125^{\circ} \mathrm{C}$; $v_{\text {max. }} 1730$ and $1715 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.93$, 1.06, and 1.13 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $0.90-2.57(19 \mathrm{H}, \mathrm{m}$, $8 \times \mathrm{CH}_{2}$ and $3 \times \mathrm{CH}$ ), and $3.69(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}) ; m / z 332\left(M^{+}\right)$ (Found: $M^{+}, 332.2352 . \mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{3}$ requires $M, 332.2349$ ).
( $\pm$ )-(3RS,4aRS,4bSR,8aRS,10aRS)-3,4,4a,4b,5,6,7,8,8a,9,10,-10a-Dodecahydro-4b,8,8-trimethyl-3,10a-ethanophenanthren$1(2 \mathrm{H})$-one (25).-To a solution of the keto ester (23) $(8 \mathrm{mg}$, 0.024 mmol ) in dry diethyl ether ( 1 ml ) at $-78^{\circ} \mathrm{C}$ was slowly added DIBAL ( $13.6 \mathrm{mg}, 0.096 \mathrm{mmol}$ ). After attaining room temperature, the mixture was stirred for 2 h at the same temperature. After addition of water ( 0.05 ml ), the mixture was further stirred for 30 min at room temperature and then filtered through Celite. Evaporation of the filtrate gave a residue, which was dissolved in DMF ( 1.5 ml ). After addition of PDC ( 22 mg , 0.06 mmol ), the mixture was stirred for 2 h at room temperature and then diluted with water ( 5 ml ). The mixture was extracted with benzene and the extract was washed with water, dried, and evaporated. Purification of the product by silica gel column chromatography with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) as eluant yielded the aldehyde (24) ( $4 \mathrm{mg}, 55 \%$ ) as needles, m.p. $121-123^{\circ} \mathrm{C} ; v_{\text {max. }} 1720 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.95,1.07$, and 1.13 (each 3 H , each s, $3 \times \mathrm{Me}$ ) and $9.73(1 \mathrm{H}$, br s, CHO); $m / z$ 302 ( $M^{+}$).

To a solution of the above product (24) ( $4 \mathrm{mg}, 0.013 \mathrm{mmol})$ in dry xylene ( 1 ml ) was added TTPRCl ( $15 \mathrm{mg}, 0.016 \mathrm{mmol}$ ) and the mixture was stirred and heated for 1 h at $140^{\circ} \mathrm{C}$. After evaporation of the solvent, the residue was taken up into diethyl ether and the solution was filtered through Celite. Evaporation of the filtrate afforded a residue, which was chromatographed on silica gel. Elution with $n$-hexane-ethyl acetate ( $10: 1$, v/v) gave the ketone (25) ( $1 \mathrm{mg}, 28 \%$ ) as needles, m.p. $80-82^{\circ} \mathrm{C}$; $v_{\text {max. }} 1710 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.91,0.97$, and 1.14 (each 3 H , each s, $3 \times \mathrm{Me})$ and $1.10-2.40\left(21 \mathrm{H}, \mathrm{m}, 9 \times \mathrm{CH}_{2}\right.$ and $\left.3 \times \mathrm{CH}\right)$; $m / z 274\left(M^{+}\right)$(Found: $M^{+}, 274.2296 . \mathrm{C}_{19} \mathrm{H}_{30} \mathrm{O}$ requires $M$, 274.2301).
( $\pm$ )-(2RS,3RS,4aRS,4bSR,8aRS,10RS)-3,4,4a,4b,5,6,7,8,8a,-9,10,10a-Dodecahydro-2,4b,8,8-tetramethyl-3,10a-ethano-phenanthren-1 2 H )-one (26).-To a stirred solution of LDA, prepared from di-isopropylamine ( $3.6 \mathrm{mg}, 0.036 \mathrm{mmol}$ ) and butyl-lithium ( $2.3 \mathrm{mg}, 0.036 \mathrm{mmol}$ ), in dry THF ( 0.5 ml ) at $-78^{\circ} \mathrm{C}$ was added a solution of the ketone (25) ( $2 \mathrm{mg}, 0.007$ mmol ) in dry THF ( 0.5 ml ), and the mixture was stirred for 2 h at $-78^{\circ} \mathrm{C}$ under argon. After addition of methyl iodide $(7.2 \mathrm{mg}$, 0.05 mmol ), the mixture was allowed to attain $0^{\circ} \mathrm{C}$ and was then stirred for 3 h at this temperature. After addition of saturated aqueous ammonium chloride ( 1 ml ), followed by extraction with diethyl ether, the extract was dried and evaporated to give a residue, which was purified by h.p.l.c. [LiChrosorb SI 60 $(4 \times 250 \mathrm{~mm})$; n-hexane-ethyl acetate ( $100: 3, \mathrm{v} / \mathrm{v}$ ) flow rate 2.0 $\mathrm{ml} \mathrm{min}^{-1}$ ] to afford the tetramethyl compound ( 26 ) $(1.4 \mathrm{mg}, 67 \%$ ) as an oil, $v_{\text {max. }} 1710 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.91,0.99$, and 1.14 (each 3 H , each s, $3 \times \mathrm{Me}$ ) and $1.17(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 2-\mathrm{Me}) ; m / z 288$ $\left(M^{+}\right)$(Found: $M^{+}, 288.2451 . \mathrm{C}_{20} \mathrm{H}_{32} \mathrm{O}$ requires $M, 288.2450$ ).
(+)-(4aS,5S)-4,4a,5,6,7,8-Hexahydro-5-hydroxy-4a-methyl-naphthalen- $2(3 \mathrm{H}$ )-one (28).-To a solution of the enone (27) (47 $\mathrm{g}, 0.264 \mathrm{~mol})$ in ethanol (11) at $0^{\circ} \mathrm{C}$ was added portionwise sodium borohydride ( $3.3 \mathrm{~g}, 0.087 \mathrm{mmol}$ ) and the mixture was
stirred for 2 min at $0^{\circ} \mathrm{C}$, then treated with acetic acid ( 30 ml ), stirring for another 5 min at $0^{\circ} \mathrm{C}$ and then evaporated. The residue was partitioned between methylene dichloride and saturated aqueous sodium chloride. The organic layer was dried and evaporated to give a residue, which was subjected to silica gel column chromatography. Elution with n-hexane-ethyl acetate ( $1: 1, \mathrm{v} / \mathrm{v}$ ) afforded the alcohol $(\mathbf{2 8})(45.15 \mathrm{~g}, 95 \%)$ as an oil, $[\alpha]_{\mathrm{D}}^{15}+122.2^{\circ}(c \quad 0.18)$ (Found: C, 73.4; H, 9.25. $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}_{2}$ requires C, $73.3 ; \mathrm{H}, 8.95 \%$ ); $v_{\text {max. }} 3600(\mathrm{OH})$ and $1660 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 1.13(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.25-2.50(10 \mathrm{H}, \mathrm{m}$, $\left.5 \times \mathrm{CH}_{2}\right), 3.05(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.06-3.65(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H})$, and $5.63(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 1-\mathrm{H}) ; \mathrm{m} / \mathrm{z} 180\left(\mathrm{M}^{+}\right)$.
(+)-(4aS,5S)-5-Benzoyloxy-4,4a,5,6,7,8-hexahydro-4a-methylnaphthalen-2(3H)-one (29).-To a solution of the alcohol (28) ( $55.0 \mathrm{~g}, 0.306 \mathrm{~mol})$ in pyridine $(26 \mathrm{ml}, 0.322 \mathrm{~mol})$ at $0^{\circ} \mathrm{C}$ was added benzoyl chloride ( $37 \mathrm{ml}, 0.322 \mathrm{~mol}$ ), and the mixture was stirred for 10 h at room temperature. After filtration, the filtrate was diluted with water and then extracted with diethyl ether. The extract was washed with water, dried, and evaporated to give a residue, which was purified by silica gel column chromatography. Elution with n-hexane-ethyl acetate (5:1, $\mathrm{v} / \mathrm{v}$ ) yielded the benzoate (29) ( $83.31 \mathrm{~g}, 96 \%$ ) as an oil, $[\alpha]_{\mathrm{D}}{ }^{5}$ $+158.9^{\circ}(c 0.18)$ (Found: C, 76.2; H, 7.2. $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{O}_{3}$ requires $\mathrm{C}, 76.05 ; \mathrm{H}, 7.1 \%$ ); $v_{\text {max. }} 1715$ and $1660 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 1.43$ $(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.40-2.50\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right), 4.89(1 \mathrm{H}, \mathrm{dd}, J$ 4.3 and $10 \mathrm{~Hz}, 5-\mathrm{H}), 5.83(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 1-\mathrm{H}), 7.31-7.60(3 \mathrm{H}, \mathrm{m}$, ArH ), and 8.91-9.07 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ); $m / z 284\left(M^{+}\right)$.
(+)-(4aS,5S)-5-Benzoyloxy-3,4,4a,5,6,7-hexahydro-1,1,4a-tri-methylnaphthalen- $2(1 \mathrm{H}$ )-one $(\mathbf{3 0})$.-To a stirred solution of potassium $t$-butoxide in $t$-butyl alcohol, prepared from potassium ( $570 \mathrm{mg}, 21.1 \mathrm{mmol}$ ) and t-butyl alcohol ( 30 ml ), was added a solution of the enone ( 29$)(2.01 \mathrm{~g}, 7.1 \mathrm{mmol})$ in t -butyl alcohol $(10 \mathrm{ml})$, and the mixture was stirred for 15 min at room temperature. After addition of methyl iodide ( $1.54 \mathrm{ml}, 24.9$ mmol ), the resulting mixture was further stirred for 45 min at ambient temperature and was then neutralised at $0^{\circ} \mathrm{C}$ by careful addition of $10 \%$ hydrochloric acid. The mixture was extracted with diethyl ether and the extract was washed with saturated aqueous sodium chloride, and dried. Evaporation of the solvent gave a residue, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $100: 3$, v/v) afforded the ketone (30) $(1.86 \mathrm{~g}, 84 \%)$ as an oil, $[\alpha]_{\mathrm{D}}^{15}+22.4^{\circ}$ (c 0.17) (Found: C, 77.05; H, 7.5. $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{O}_{3}$ requires C, $76.9 ; \mathrm{H}$, $7.75 \%)$; $v_{\text {max. }} 1710 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 1.19,1.27$, and 1.29 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $1.30-2.60\left(8 \mathrm{H}, \mathrm{m}, 4 \times \mathrm{CH}_{2}\right), 5.04(1 \mathrm{H}, \mathrm{dd}, J$ 5.7 and $10.8 \mathrm{~Hz}, 5-\mathrm{H}), 5.60(1 \mathrm{H}, \mathrm{t}, J 4.3 \mathrm{~Hz}, 8-\mathrm{H}), 7.29-7.64(3$ $\mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ), and $7.71-8.13(2 \mathrm{H}, \mathrm{m}, \mathrm{ArH}) ; m / z 312\left(\mathrm{M}^{+}\right)$.
(-)-(1S,8aS)-1,2,3,5,6,7,8,8a-Octahydro-5,5,8a-trimethyl-
naphthalen-1-ol (31).-A mixture of the ketone (30) $(1.8 \mathrm{~g}, 5.8$ mmol ), potassium hydroxide ( $3.0 \mathrm{~g}, 46 \mathrm{mmol}$ ), and hydrazine hydrate ( $3.3 \mathrm{ml}, 68 \mathrm{mmol}$ ) in diethylene glycol ( 33 ml ) was refluxed for 2 h , and then the excess of hydrazine hydrate was evaporated off. The reaction mixture was further refluxed for 4 h , diluted with water ( 100 ml ), and then extracted with diethyl ether. The extract was dried and evaporated to give a residue, which was purified by silica gel column chromatography. Elution with $n$-hexane-ethyl acetate ( $5: 1, \mathrm{v} / \mathrm{v}$ ), followed by recrystallisation of the product from $n$-hexane afforded the alcohol (31) $(1.052 \mathrm{~g}, 93 \%)$ as needles, m.p. $82-83^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{5}$ $-58.8^{\circ}$ (c 0.16) (Found: C, 80.3; H, 11.35. $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}$ requires $\mathrm{C}, 80.35 ; \mathrm{H}, 11.4 \%$ ); $v_{\text {max. }} 3600 \mathrm{~cm}^{-1}(\mathrm{OH}) ; \delta_{\mathrm{H}} 1.06(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Me}), 1.11(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Me}), 1.20-2.25\left(10 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right)$, $3.41(1 \mathrm{H}, \mathrm{dd}, J 7.1$ and $8.6 \mathrm{~Hz}, 1-\mathrm{H})$, and $5.37(1 \mathrm{H}, \mathrm{t}, J 3.6 \mathrm{~Hz}$, 4-H); $m / z 194$ ( $M^{+}$).
(+)-(1S,4aS,8aS)-1,2,3,4,4a,5,6,7,8,8a-Decahydro-5,5,8a-tri-methylnaphthalen-1-ol (32).-A mixture of the olefin (31) ( $820 \mathrm{mg}, 4.18 \mathrm{mmol}$ ) and Adams' catalyst ( 120 mg ) in acetic acid ( 10 ml ) was stirred under an atmosphere of hydrogen until absorption of hydrogen had ceased. After filtration, followed by addition of water ( 20 ml ), the mixture was extracted with diethyl ether. The extract was washed with saturated aqueous sodium hydrogen carbonate, dried, and evaporated to give a residue, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate, followed by recrystallisation from n -hexane, afforded the alcohol ( $\mathbf{3 2}$ ) ( $792 \mathrm{mg}, 95 \%$ ) as needles, m.p. $88-89^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{17}+1.50^{\circ}$ (c 0.40 ) (Found: C , 79.45; $\mathrm{H}, 12.45 . \mathrm{C}_{13} \mathrm{H}_{24} \mathrm{O}$ requires $\mathrm{C}, 79.55 ; \mathrm{H}, 12.3 \%$ ); $\mathrm{v}_{\text {max }}$. $3610 \mathrm{~cm}^{-1}(\mathrm{OH}) ; \delta_{\mathrm{H}} 0.84,0.86$, and 0.89 (each 3 H , each s, $3 \times \mathrm{Me}), 0.80-1.90\left(13 \mathrm{H}, \mathrm{m}, 6 \times \mathrm{CH}_{2}\right.$ and CH ), and 3.14 ( 1 $\mathrm{H}, \mathrm{dd}, J 4.3$ and $10 \mathrm{~Hz}, 1-\mathrm{H}$ ); $m / z 196$ ( $M^{+}$).

> (-)-(4aS,8aS)-3,4,4a,5,6,7,8,8a-Octahydro-5,5,8a-trimethyl-
naphthalen-1 (2H)-one (33).-To a solution of the alchohol (32) ( $792 \mathrm{mg}, 4 \mathrm{mmol}$ ) in methylene dichloride ( 20 ml ) were added Florisil ( 1.3 g ) and PCC ( $1.3 \mathrm{~g}, 6 \mathrm{mmol}$ ), and the mixture was stirred for 2 h at room temperature. After filtration through Celite, followed by evaporation of the filtrate, the residue was subjected to silica gel column chromatography. Elution with n -hexane-ethyl acetate ( $100: 5, \mathrm{v} / \mathrm{v}$ ) yielded the ketone (33) (706 $\mathrm{mg}, 90 \%$ ) as an oil, $[x]_{\mathrm{D}}^{17}-39.1^{\circ}(c 0.44)$; $v_{\text {max }} 1700 \mathrm{~cm}^{-1}$ ( $\mathrm{C}=\mathrm{O}$ ); $\delta_{\mathrm{H}} 0.89,0.96$, and 1.15 (each 3 H , each s, $3 \times \mathrm{Me}$ ); $m / z$ $194\left(M^{+}\right)$(Found: $M^{+}$, 194.1668. $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}$ requires $M$, 194.1667).
(-)-(2R,4aS,8aS)-2-Bromo-3,4,4a,5,6,7,8,8a-octahydro-5,5,8a-trimethylnaphthalen- $1(2 \mathrm{H}$ )-one (34).-To a solution of the ketone ( $\mathbf{3 3}$ ) ( $62 \mathrm{mg}, 0.32 \mathrm{mmol}$ ) in acetic acid ( 5 ml ) was added PBB ( $101 \mathrm{mg}, 0.32 \mathrm{mmol}$ ), and the mixture was stirred for 2 h at room temperature. After addition of water ( 10 ml ), the resulting mixture was extracted with diethyl ether. The extract was washed with saturated aqueous sodium hydrogen carbonate, dried, and evaporated to give a residue, which was chromatographed on silica gel and eluted with n-hexane-ethyl acetate ( $100: 2, \mathrm{v} / \mathrm{v}$ ). Recrystallisation of the product from n -hexane afforded the bromide (34) $(84 \mathrm{mg}, 97 \%)$ as needles, $\mathrm{m} . \mathrm{p}$. $126-128^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{17}-22.0^{\circ}(c 0.30$ ) (Found: C, $57.15 ; \mathrm{H}, 7.55$; Br 29.3. $\mathrm{C}_{13} \mathrm{H}_{21} \mathrm{BrO}$ requires C, $57.15 ; \mathrm{H}, 7.7$; $\mathrm{Br}, 29.3 \%$ ); $\mathrm{v}_{\text {max }}$. $1720 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.90,0.91$, and 1.17 (each 3 H , each s , $3 \times \mathrm{Me}), 1.15-2.80\left(11 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}\right.$ and CH$)$, and $4.94(1$ H, dd, $J 5.7$ and $11.4 \mathrm{~Hz}, 2-\mathrm{H}$ ); m/z $272\left(M^{+}\right)$.

> (-)-(4aS,8aS)-3,4,4a,5,6,7,8,8a-Octahydro-2-hydroxy-5,5,8- trimethylnaphthalen- $1(2 \mathrm{H}$ )-one (35)--To a solution of the bromide (34) ( $40 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) in DMF ( 3 ml ) at room temperature was slowly added $0.74 \%$ aqueous sodium hydroxide ( 1 ml ), and the mixture was stirred for 30 min at the same temperature before being neutralised at $0^{\circ} \mathrm{C}$ by addition of $1 \%$ hydrochloric acid. The mixture was extracted with diethyl ether, the extract was washed with saturated aqueous sodium chloride and dried, and evaporation of the solvent afforded a residue, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) yielded the acyloin (35) ( $30 \mathrm{mg}, 97 \%$ ) as an oil, $[\alpha]_{\mathrm{D}}^{17}-28.9^{\circ}$ (c 0.28 ) (Found: C, 74.2; H, 10.4. $\mathrm{C}_{13} \mathrm{H}_{22} \mathrm{O}_{2}$ requires $\mathrm{C}, 74.25 ; \mathrm{H}$, $10.55 \%) ; v_{\text {max }} .3470(\mathrm{OH})$ and $1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.91$, 0.93 , and 1.16 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $1.05-2.58(11 \mathrm{H}, \mathrm{m}$, $5 \times \mathrm{CH}_{2}$ and CH$), 4.60(1 \mathrm{H}, \mathrm{d}, J 3.4 \mathrm{~Hz}, \mathrm{OH})$, and $4.38(1 \mathrm{H}$, ddd, $J 3.4,7.1$, and $11.4 \mathrm{~Hz}, 2-\mathrm{H}) ; m / z 210\left(M^{+}\right)$.

> (-)-(1S,2S)-2-(3-Hydroxypropyl)-1,3,3-trimethylcyclo-
hexanecarboxylate (36).-To a solution of the acyloin (35) (125 $\mathrm{mg}, 0.6 \mathrm{mmol}$ ) in a mixture of methanol ( 1 ml ) and benzene ( 3
ml) at $0^{\circ} \mathrm{C}$ was added LTA ( $293 \mathrm{mg}, 0.6 \mathrm{mmol}$ ), and the mixture was stirred for 10 min at room temperature. After addition of saturated aqueous sodium hydrogen carbonate, followed by filtration through Celite, the filtrate was extracted with benzene. The extract was dried and evaporated to give the crude aldehyde ( $128 \mathrm{mg}, 90 \%$ ), which was used in the next reaction without further purification.
To a solution of the above product ( $40 \mathrm{mg}, 0.17 \mathrm{mmol}$ ) in methanol at $0^{\circ} \mathrm{C}$ was added sodium borohydride ( $6 \mathrm{mg}, 0.17$ mmol ), and the mixture was stirred for 20 min at room temperature. After evaporation of the solvent, the residue was partitioned between saturated aqueous sodium chloride and methylene dichloride. The organic layer was dried and evaporated to give a residue, which was subjected to silica gel column chromatography. Elution with n-hexane-ethyl acetate ( $10: 4, \mathrm{v} / \mathrm{v}$ ) afforded the alcohol ( $\mathbf{3 6}$ ) ( $39 \mathrm{mg}, 97 \%$ ) as an oil, $[\alpha]_{\mathrm{D}}{ }^{7}-5.70^{\circ}(c 0.14) ; v_{\text {max. }} 3600(\mathrm{OH})$ and $1715 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O})$; $\delta_{\mathrm{H}} 0.89,0.93$, and $1.19($ mach. 3 H , each s, $3 \times \mathrm{Me}$ ), $1.10-1.85(11$ $\mathrm{H}, \mathrm{m}, 5 \times \mathrm{CH}_{2}$ and CH ), $3.56\left(2 \mathrm{H}\right.$, br t, $J 5.7 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{O}$ ), and $3.64(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$; m/z $242\left(M^{+}\right)$(Found: $M^{+}$, 242.1882. $\mathrm{C}_{14} \mathrm{H}_{26} \mathrm{O}_{3}$ requires $M, 242.1884$ ).
(-)-(1S,2S)-1,3,3-Trimethyl-2-[3-(2-nitrophenylseleno)propyl]cyclohexanecarboxylate (37).-To a stirred solution of the alcohol ( $\mathbf{3 6}$ ) ( $25 \mathrm{mg}, 0.1 \mathrm{mmol}$ ) and 2-nitrophenylselenocyanate ( $47 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) in dry THF ( 3 ml ) was added tributylphosphine ( $0.05 \mathrm{ml}, 0.2 \mathrm{mmol}$ ) at room temperature, and the mixture was stirred for 20 min at the same temperature. Evaporation of the solvent, followed by purification of the residue by silica gel chromatography with n-hexane-ethyl acetate ( $100: 5, \mathrm{v} / \mathrm{v}$ ) as eluant, gave the selenide ( $\mathbf{3 7}$ ) ( $48 \mathrm{mg}, 98 \%$ ) as a yellow oil, $[\alpha]_{\mathrm{D}}^{17}-3.30^{\circ}(c 0.18)$; $v_{\text {max }} 1710 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O})$; $\delta_{\mathrm{H}} 0.91(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Me}), 1.19(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 1.25-2.10(11 \mathrm{H}, \mathrm{m}$, $5 \times \mathrm{CH}_{2}$ and CH ), $2.85\left(2 \mathrm{H}, \mathrm{brt}, J 6 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Se}\right), 3.57(3 \mathrm{H}, \mathrm{s}$, $\mathrm{OMe}), 7.15-7.57(3 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$, and $8.22(1 \mathrm{H}, \mathrm{brd}, J 7 \mathrm{~Hz}$, ArH); $m / z 425\left(M^{+}\right)$and $427\left(M^{+}+2\right.$ ) [Found: $M^{+}, 425.1269$ and $(M+2)^{+}$427.1262. $\mathrm{C}_{20} \mathrm{H}_{29} \mathrm{NO}_{4}$ Se requires $M, 425.1304$ and $(M+2), 427.1280]$.

Methyl ( + )-(1S,2S)-2-Allyl-1,3,3-trimethylcyclohexanecarboxylate (38).-To a stirred solution of the selenide (37) (40 mg ), 0.094 mmol ) in THF ( 3 ml ) at $0^{\circ} \mathrm{C}$ was added $30 \%$ aqueous hydrogen peroxide ( 0.1 ml ), and the mixture was stirred for 10 h at room temperature. After addition of water ( 5 $\mathrm{ml})$, the mixture was extracted with diethyl ether. The extract was washed with saturated aqueous sodium chloride, dried, and evaporated to give a residue, which was purified by chromatography on silica gel. Elution with n-hexane-ethyl acetate ( $100: 5, \mathrm{v} / \mathrm{v}$ ) afforded the olefin ( $\mathbf{3 8}$ ) ( $19 \mathrm{mg}, 91 \%$ ) as an oil, $[\alpha]]^{17}+22.9^{\circ}(c 0.28)$; $v_{\text {max }} 1710$ and $1640 \mathrm{~cm}^{-1}$ (C=O); $\delta_{\mathrm{H}} 0.89,0.94$, and $1.20($ each 3 H , each s, $3 \times \mathrm{Me}), 1.20-2.30(9$ $\mathrm{H}, \mathrm{m}, 4 \times \mathrm{CH}_{2}$ and CH ), $3.54(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 4.77(1 \mathrm{H}, \mathrm{brd}, J 10$ $\mathrm{Hz}, \mathrm{CH}=\mathrm{C} H \mathrm{H}), 4.82(1 \mathrm{H}, \mathrm{brd}, J 16 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CH} H)$, and $5.40-$ $5.87\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CH}_{2}\right) ; \mathrm{m} / \mathrm{z} 224\left(M^{+}\right)$(Found: $M^{+}$, 224.1777. $\mathrm{C}_{14} \mathrm{H}_{24} \mathrm{O}_{2}$ requires $M, 244.1777$ ).

## (+)-(1S,2S)-(2-Allyl-1,3,3-trimethylcyclohexyl)methanol

(39).-To a stirred mixture of lithium aluminium hydride ( 34 $\mathrm{mg}, 0.89 \mathrm{mmol}$ ) in dry diethyl ether ( 5 ml ) at $0^{\circ} \mathrm{C}$ was added dropwise a solution of the ester ( $\mathbf{3 8}$ ) ( $100 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) in dry diethyl ether ( 1 ml ), and the mixture was stirred for 30 min at room temperature. After successive addition of water ( 0.03 ml ), $15 \%$ aqueous sodium hydroxide ( 0.03 ml ), and water ( 0.09 ml ), followed by stirring for 30 min , the mixture was filtered through Celite and washed with diethyl ether. Evaporation of the combined filtrate and washings gave a residue, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) afforded a solid, which was recrystallised
from $n$-hexane to yield the alcohol ( 39 ) ( $83 \mathrm{mg}, 95 \%$ ) as needles, m.p. $30-31^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{17}+5.0^{\circ}(c \quad 0.24)$; $v_{\text {max. }} 3550(\mathrm{OH})$ and $1620 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{C}) ; \delta_{\mathrm{H}} 0.80,0.89$, and 0.93 (each 3 H , each s, $3 \times \mathrm{Me}), 1.15-1.60\left(7 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{CH}_{2}\right.$ and CH$), 1.91-2.14$ ( 2 $\mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}=$ ), 2.97 and 3.43 (each 1 H , each br d, $J 11.4 \mathrm{~Hz}$, $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 4.83(1 \mathrm{H}$, br d, $J 10 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CHH}), 4.95(1 \mathrm{H}, \mathrm{brd}, J$ $16 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CH} H)$, and $5.47-6.20\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CH}_{2}\right) ; m / z 196$ ( $M^{+}$) (Found: $M^{+}, 196.1827 . \mathrm{C}_{13} \mathrm{H}_{24} \mathrm{O}$ requires $M, 196.1829$ ).

## (+)-(1S,2S)-2-Allyl-1,3,3-trimethylcyclohexanecarbaldehyde

 Ethylene Acetal (40).-To a stirred solution of the alcohol (39) $(100 \mathrm{mg}, 0.51 \mathrm{mmol})$ in dry methylene dichloride ( 3 ml ) were added Florisil ( 220 mg ) and PCC ( $220 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) at room temperature, and the mixture was stirred for 1 h at the same temperature. After filtration through Celite, followed by evaporation of the filtrate, the residue was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $100: 3, \mathrm{v} / \mathrm{v}$ ) gave the corresponding aldehyde ( $87 \mathrm{mg}, 88 \%$ ) as an oil, $v_{\text {max. }} 1710$ $\mathrm{cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.93,1.00$, and 1.10 (each 3 H , each s, $3 \times \mathrm{Me}), 4.87(1 \mathrm{H}$, br d, $J 16 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CHH}), 4.88(1 \mathrm{H}$, br d, $J$ $10 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CH} H), 5.35-5.90\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CH}_{2}\right)$, and $9.07(1$ H, s, CHO); $m / z 194$ ( $M^{+}$).A mixture of the above aldehyde ( $30 \mathrm{mg}, 0.15 \mathrm{mmol}$ ), ethylene glycol ( $0.013 \mathrm{ml}, 0.23 \mathrm{mmol}$ ), and toluene-p-sulphonic acid (3 mg ) in dry benzene ( 5 ml ) was refluxed for 20 min in a DeanStark apparatus. After dilution with benzene, the mixture was washed successively with saturated aqueous sodium hydrogen carbonate and water, dried, and evaporated. Purification of the crude product by silica gel column chromatography with n -hexane-ethyl acetate ( $40: 1, \mathrm{v} / \mathrm{v}$ ) as eluant yielded the acetal (40) ( $36 \mathrm{mg}, 98 \%$ ) as an oil, $[\alpha]_{\mathrm{D}}^{17}+5.0^{\circ}(c 0.08)$; $v_{\text {max. }} 1630$ $\mathrm{cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.87(6 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Me}), 0.90(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$, $1.10-1.50\left(7 \mathrm{H}, \mathrm{m}, 3 \times \mathrm{CH}_{2}\right.$ and CH$), 2.00-2.40(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}_{2} \mathrm{CH}=\right)$, $3.63-3.90\left(4 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 4.50(1 \mathrm{H}, \mathrm{s}$, OCHO), $4.73(1 \mathrm{H}$, br d, $J 10 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CH} H), 4.82(1 \mathrm{H}$, br d, $J$ $16 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CHH}$ ), and $5.40-6.10\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CH}_{2}\right)$ ) $m / z 238$ ( $M^{+}$) (Found: $M^{+}, 238.1931 . \mathrm{C}_{15} \mathrm{H}_{26} \mathrm{O}_{2}$ requires $M, 238.1914$ ).
$(-)-\left(1^{\prime \prime} \mathrm{S}, 2^{\prime \prime} \mathrm{S}\right)-(E)-2-\left\{2^{\prime}-\left[2^{\prime \prime}-(\right.\right.$ Dioxolan $-2-y l)-2^{\prime \prime}, 6^{\prime \prime}, 6^{\prime \prime}-$ trimethylcyclohexyl] ethylidene $\}$ cyclohexanone (42).-A mixture of the olefin ( 40 ) ( $159 \mathrm{mg}, 0.67 \mathrm{mmol}$ ), osmium tetraoxide ( 8 $\mathrm{mg}, 0.003 \mathrm{mmol})$, diethyl ether ( 6 ml ), and water ( 6 ml ) was stirred for 5 min at room temperature, sodium periodate (4.29 $\mathrm{g}, 20 \mathrm{mmol}$ ) was addded, and the mixture was stirred for 10 h at the same temperature and then extracted with diethyl ether. The extract was washed with saturated aqueous sodium chloride, dried, and evaporated to give a residue, which was subjected to chromatography on silica gel. Elution with $n$-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) afforded the corresponding aldehyde (41) ( $125 \mathrm{mg}, 78 \%$ ) as an oil, $v_{\text {max. }} 1720 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O})$; $\delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.82,0.85$, and 0.90 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $3.60-$ $3.90\left(4 \mathrm{H}, \mathrm{m} \mathrm{OCH} \mathrm{CH}_{2} \mathrm{O}\right), 4.20(1 \mathrm{H}, \mathrm{s}, \mathrm{OCHO})$, and $9.52(1$ $\mathrm{H}, \mathrm{t}, J 1 \mathrm{~Hz}, \mathrm{CHO}) ; m / z 240\left(M^{+}\right)$.
To a stirred solution of LDA, prepared from di-isopropylamine ( $37 \mathrm{mg}, 0.37 \mathrm{mmol}$ ) and butyl-lithium ( $23 \mathrm{mg}, 0.37$ mmol ), in dry THF ( 3 ml ) at $-78^{\circ} \mathrm{C}$ was slowly added cyclohexanone ( $55 \mathrm{mg}, 0.23 \mathrm{mmol}$ ), and the mixture was stirred for 1 h at this temperature. After slow addition, at $-78^{\circ} \mathrm{C}$, of a solution of the above aldehyde (41) ( $55 \mathrm{mg}, 0.23 \mathrm{mmol}$ ) in dry THF ( 1 ml ), the mixture was allowed to warm to room temperature and was stirred for 1 h at the same temperature. After addition, at $0^{\circ} \mathrm{C}$, of saturated aqueous ammonium chloride, the mixture was extracted with diethyl ether. The extract was dried and evaporated to give a residue, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $25: 2, \mathrm{v} / \mathrm{v}$ ) afforded a solid, which was recrystallised from n-hexane to afford the enone (42) ( $45 \mathrm{mg}, 62 \%$ ) as needles, m.p. $52-53^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}^{17}-10.0^{\circ}$ (c 0.08); $v_{\text {max. }} 1670 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O})$;
$\delta_{H} 0.84,0.90$, and 0.97 (each 3 H , each s, $3 \times \mathrm{Me}$ ), 1.05-2.60(17 $\mathrm{H}, \mathrm{m}, 8 \times \mathrm{CH}_{2}$ and CH$), 3.66-3.91\left(4 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $4.43(1 \mathrm{H}, \mathrm{s}, \mathrm{OCHO})$, and $6.69(1 \mathrm{H}, \mathrm{tt}, J 1.4$ and 6.6 Hz , olefinic H ); $m / z 320\left(M^{+}\right)$(Found: $M^{+}, 320.2349 . \mathrm{C}_{20} \mathrm{H}_{32} \mathrm{O}_{3}$ requires $M, 320.2349)$.

Methyl (1'S,2'S)-3-\{1', $3^{\prime}, 3^{\prime}$-Trimethyl-2'-[2-(2-oxocyclohex-3-enyl) ethyl] cyclohexyl \}prop-2( E )-enoate (46).-A mixture of the enone ( 42 ) ( $120 \mathrm{mg}, 0.37 \mathrm{mmol}$ ) and $10 \%$ palladiumcharcoal ( 50 mg ) in ethanol ( 10 ml ) was stirred under an atmosphere of hydrogen until absorption of hydrogen had ceased. After filtration, the filtrate was evaporated and the residue was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $100: 8, \mathrm{v} / \mathrm{v}$ ) yielded the ketone (43) (120 $\mathrm{mg}, 98 \%$ ) as an oil, $v_{\text {max. }} 1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.87(6 \mathrm{H}$, $\mathrm{s}, 2 \times \mathrm{Me}), 0.90(3 \mathrm{H}, \mathrm{s}, 3 \times \mathrm{Me}), 3.68-3.80(4 \mathrm{H}, \mathrm{m}$, $\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}$ ), and 4.43 and 4.56 [1 H (1:2), each s, OCHO]; $m / z 322\left(M^{+}\right)$.

To a stirred solution of LDA, prepared from di-isopropylamine ( $16 \mathrm{mg}, 0.186 \mathrm{mmol}$ ) and butyl-lithium ( $10 \mathrm{mg}, 0.186$ mmol ), in dry THF ( 1 ml ) at $-78^{\circ} \mathrm{C}$ under argon was added a solution of the ketone (43) ( $50 \mathrm{mg}, 0.155 \mathrm{mmol}$ ) in dry THF ( 0.5 ml ). After being stirred for 1 h at $-78^{\circ} \mathrm{C}$, the mixture was treated slowly with a solution of trimethylsilyl chloride ( 34 mg , 0.31 mmol ) and triethylamine ( $7 \mathrm{mg}, 0.07 \mathrm{mmol}$ ) in dry THF $(0.5 \mathrm{ml})$. The resulting mixture was allowed to warm to room temperature and was then stirred for a further 1 h . After addition, at $0^{\circ} \mathrm{C}$, saturated aqueous sodium hydrogen carbonate ( 3 ml ), the mixture was extracted with diethyl ether. The extract was dried and evaporated to give a residue, which was dissolved in acetonitrile ( 3 ml ). After addition of palladium(II) acetate ( $52 \mathrm{mg}, 0.23 \mathrm{mmol}$ ) and $p$-benzoquinone ( $17 \mathrm{mg}, 0.155$ mmol ), the resulting mixture was stirred overnight at room temperature under nitrogen. After evaporation of the solvents, the residue was taken up into benzene and the solution was filtered. Concentration of the filtrate afforded a residue, which was chromatographed on silica gel. Elution with n -hexaneethyl acetate ( $100: 5, \mathrm{v} / \mathrm{v}$ ) yielded the enone (44) ( $42 \mathrm{mg}, 85 \%$ ) as an oil, $v_{\text {max. }} 1670 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.88(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}), 0.90(6$ $\mathrm{H}, \mathrm{s}, 2 \times \mathrm{Me}), 3.73-3.95\left(4 \mathrm{H}, \mathrm{m}, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 4.31$ and 4.35 [ $1 \mathrm{H}(1: 2)$, each s, OCHO], $5.85(1 \mathrm{H}$, br d, $J 10 \mathrm{~Hz}, \mathrm{COCH}=)$, and 6.55-6.97 ( $1 \mathrm{H}, \mathrm{m},=\mathrm{CHCH}_{2}$ ); $m / z 320\left(\mathrm{M}^{+}\right)$.

A mixture of the enone (44) ( $42 \mathrm{mg}, 0.13 \mathrm{mmol}$ ) and $10 \%$ perchloric acid ( 3 ml ) in THF ( 3 ml ) was stirred for 1 h at room temperature. After addition of saturated aqueous sodium hydrogen carbonate, the mixture was extracted with diethyl ether. The extract was washed with water, dried, and evaporated to give a residue, which was subjected to silica gel column chromatography. Elution with n-hexane-ethyl acetate (10:1, $\mathrm{v} / \mathrm{v}$ ) gave the aldehyde ( $\mathbf{4 5 ) ( 3 0 \mathrm { mg } , 8 3 \% ) \text { as an oil, } v _ { \text { max. } } 1 7 1 0}$ and $1665 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.92,1.00$, and 1.05 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $5.83(1 \mathrm{H}$, br d, $J 10 \mathrm{~Hz}, \mathrm{COCH}=), 6.57-6.98(1$ $\mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}=$ ), and $9.18(1 \mathrm{H}, \mathrm{s}, \mathrm{CHO}) ; m / z 276\left(\mathrm{M}^{+}\right)$.

To a solution of the Emmons reagent, prepared from $60 \%$ sodium hydride ( $13 \mathrm{mg}, 0.326 \mathrm{mmol}$ ) and methyl (dimethoxyphosphoryl)acetate ( $59 \mathrm{mg}, 0.326 \mathrm{mmol}$ ) in dry DME ( 2 ml ), was added a solution of the above aldehyde (45) ( $30 \mathrm{mg}, 0.109$ mmol ) in dry DME ( 0.5 ml ), and the mixture was stirred for 72 h at room temperature under argon. After addition of water, the mixture was extracted with diethyl ether. The extract was dried and evaporated to give a residue, which was purified by silica gel column chromatography. Elution with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) afforded the $\alpha, \beta$-unsaturated ester ( 46 ) ( $17 \mathrm{mg}, 47 \%$ ) as an oil (Found: C, 75.7; H, 9.6. $\mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{3}$ requires C, $75.85 ; \mathrm{H}$, $9.7 \%) ; v_{\text {max. }} 1710$ and $1670 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right) 0.92,0.95$, and 1.05 (each 3 H , each s, $3 \times \mathrm{Me}$ ), 1.05-2.45 $(16 \mathrm{H}$, m, $7 \times \mathrm{CH}_{2}$ and $\left.2 \times \mathrm{CH}\right), 3.67(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 5.53(1 \mathrm{H}, \mathrm{d}, J 16 \mathrm{~Hz}$, $\left.=\mathrm{CHCO}_{2} \mathrm{Me}\right), 5.78(1 \mathrm{H}$, br d$, J 10 \mathrm{~Hz}, \mathrm{COCH}=), 6.53-6.95(1$
$\mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}=$ ), and $6.72\left(1 \mathrm{H}, \mathrm{d}, \mathrm{J} 16 \mathrm{~Hz}, \mathrm{CH}=\mathrm{CHCO}_{2} \mathrm{Me}\right)$; $m / z 332\left(M^{+}\right)$(Found: $M^{+}, 332.2351 . \mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{3}$ requires $M$, 332.2349).

Methyl ( - )-(3S,4R,4aR,4bS,8aS,10aR)-1,2,3,4,4a,4b,5,6,7,8,-8a,9,10,10a-Tetradecahydro-4b,8,8-trimethyl-1-oxo-3,10a-ethanophenanthrene-4-carboxylate (48).- To a stirred solution of LiHMDS, prepared from 1,1,1,3,3,3-hexamethyldisilazane ( $160 \mathrm{mg}, 0.991 \mathrm{mmol}$ ) and butyl-lithium ( $63 \mathrm{mg}, 0.991 \mathrm{mmol}$ ), in dry n -hexane ( 20 ml ) at $-78^{\circ} \mathrm{C}$ under argon was added a solution of the $\alpha, \beta$-unsaturated ester ( 46 ) $(250 \mathrm{mg}, 0.762$ mmol ) in dry diethyl ether ( 3.3 ml ). After being stirred for 1 h at $-78^{\circ} \mathrm{C}$, the mixture was allowed to warm to room temperature during 30 min , and was then stirred for 1 h . The reaction mixture was poured onto silica gel ( 50 g ) at room temperature and diethyl ether ( 100 ml ) was added. The resulting mixture was filtered and the filtrate was evaporated to give a solid, which was purified by silica gel column chromatography. Elution with n -hexane-ethyl acetate ( $20: 3, \mathrm{v} / \mathrm{v}$ ) afforded the tetracyclic compound (48) ( $230 \mathrm{mg}, 92 \%$ ) as needles, m.p. $145-148{ }^{\circ} \mathrm{C}$; $[x]_{\mathrm{D}}^{25}-9.3^{\circ}(c \quad 0.29)$ (Found: C, 75.8; H, 9.6. $\mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{3}$ requires $\mathrm{C}, 75.85 ; \mathrm{H}, 9.7 \%)$; $v_{\text {max. }} 1725$ and $1715 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}$ $0.82,0.88$, and 1.02 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $0.90-2.65(19 \mathrm{H}$, $\mathrm{m}, 8 \times \mathrm{CH}_{2}$ and $3 \times \mathrm{CH}$ ), and $3.61(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}) ; m / z 332$ ( $M^{+}$) (Found: $M^{+}, 332.2350 . \mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{3}$ requires $M, 332.2349$ ).
( - )-(3S,4R,4aR,4bS,8aS,10aR)-1,2,3,4,4a,4b,5,6,7,8,8a,9,10,-10a-Tetradecahydro-4b,8,8-trimethyl-1-oxo-3,10a-ethano-phenanthrene-4-carbaldehyde (49).-To a solution of the ester (48) ( $100 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) in dry methylene dichloride ( 7 ml ) at $-78^{\circ} \mathrm{C}$ under argon was added DIBAL ( $214 \mathrm{mg}, 1.5 \mathrm{mmol}$ ); the mixture was allowed to warm to room temperature and was then stirred for 0.5 h . After dropwise addition of water ( 2.0 ml ) to the stirred mixture at $0^{\circ} \mathrm{C}$, the resulting mixture was filtered through Celite and the filter was washed with methylene chloride. The combined filtrate and washings were washed with $2 \%$ hydrochloric acid, dried, and evaporated to give the diol ( 92 $\mathrm{mg}, 100 \%$ ) as a solid, $m / z 306\left(M^{+}\right)$(Found: $M^{+}, 306.2558$. $\mathrm{C}_{20} \mathrm{H}_{34} \mathrm{O}_{2}$ requires $M, 306.2558$ ).

A mixture of the above diol ( $92 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) and PDC $(256 \mathrm{mg}, 0.68 \mathrm{mmol})$ in dry DMF ( 5 ml ) was stirred for 3 h at room temperature, and was then treated with water $(20 \mathrm{ml})$. The resulting mixture was extracted with benzene, and the extract was washed with saturated aqueous sodium chloride, and dried. Evaporation of the solvent afforded a residue, which was chromatographed on silica gel. Elution with n-hexane-ethyl acetate ( $10: 1, \mathrm{v} / \mathrm{v}$ ) gave the aldehyde (49) ( $53.5 \mathrm{mg}, 59 \%$ ) as needles, m.p. $150-151^{\circ} \mathrm{C}$; $v_{\text {max. }} 1715 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CCl}_{4}\right)$ $0.78,0.85$, and 0.98 (each 3 H , each s, $3 \times \mathrm{Me}$ ), 1.05-2.50 (19 $\mathrm{H}, \mathrm{m}, 8 \times \mathrm{CH}_{2}$ and $3 \times \mathrm{CH}$ ), and $9.55(1 \mathrm{H}, \mathrm{s}, \mathrm{CHO}) ; m / z 302$ ( $M^{+}$) (Found: $M^{+}, 302.2246 . \mathrm{C}_{20} \mathrm{H}_{30} \mathrm{O}_{2}$ requires $M, 302.2262$ ).

## (+)-(3R,4aR,4bS,8aS,10aR)-3,4,4a,4b,5,6,7,8,8a, 9,10,10a-

Dodecahydro-4b,8,8-trimethyl-3,10a-ethanophenanthren-1(2H)one (50).-To a stirred solution of the aldehyde (49) (28 mg, 0.09 mmol ) in dry toluene ( 3 ml ) under reflux was dropwise added a solution of TTPRCl ( $172 \mathrm{mg}, 0.186 \mathrm{mmol}$ ) in dry toluene ( 3 ml ). The mixture was refluxed for 2.5 h , then the solvent was evaporated off to give a residue, which was subjected to chromatography on silica gel. Elution with benzene gave the ketone ( $\mathbf{5 0}$ ) ( $7 \mathrm{mg}, 28 \%$ ) as a solid, which on recrystallisation from benzene-n-hexane afforded crystals, m.p. $147-149^{\circ} \mathrm{C}$ (lit., ${ }^{3} 146-148{ }^{\circ} \mathrm{C}$ ); [ $\left.\alpha\right]_{\mathrm{D}}^{25}+19.5^{\circ}$ (c 0.021 in $\mathrm{MeOH})\left\{\mathrm{lit.}^{3}[\alpha]_{\mathrm{D}}+19.6^{\circ}(c 0.05\right.$ in MeOH$\left.)\right\} ; v_{\text {max. }} 1710 \mathrm{~cm}^{-1}$ $(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}} 0.83,0.86$, and 1.05 (each 3 H , each $\mathrm{s}, 3 \times \mathrm{Me}$ ) and $1.18-2.65\left(21 \mathrm{H}, \mathrm{m}, 9 \times \mathrm{CH}_{2}\right.$ and $\left.3 \times \mathrm{CH}\right) ; m / z 274\left(M^{+}\right)$ (Found: $M^{+}$, 274.2297. Calc. for $\mathrm{C}_{19} \mathrm{H}_{30} \mathrm{O}: M, 274.2310$ ); the spectral data were consistent with those reported. ${ }^{3}$
(2R,3R,4aR,4bS,8aS,10aR)-3,4,4a,4b,5,6,7,8,8a,9,10,10a-Dodecahydro- $2,4 \mathrm{~b}, 8,8$-tetramethyl-3,10-ethanophenanthren$1(2 \mathrm{H})$-one (51).-To a solution of LDA, prepared from diisopropylamine ( $1.8 \mathrm{mg}, 0.018 \mathrm{mmol}$ ) and butyl-lithium $(1.2 \mathrm{mg}$, $0.018 \mathrm{mmol})$ in dry THF ( 0.5 ml ), at $-78^{\circ} \mathrm{C}$ was slowly added a solution of the ketone ( $\mathbf{5 0}$ ) ( $1 \mathrm{mg}, 0.0036 \mathrm{mmol}$ ) in dry THF ( 0.5 ml ), and the mixture was stirred for 2 h at this temperature. After addition of methyl iodide ( $3.6 \mathrm{mg}, 0.025 \mathrm{mmol}$ ), the mixture was allowed to warm to $0^{\circ} \mathrm{C}$ and was then stirred for 30 min at this temperature. After addition of saturated aqueous ammonium chloride ( 1 ml ), the mixture was extracted with diethyl ether. The extract was dried and evaporated to give a residue, which was purified by h.p.l.c. [LiChrosorb SI $60(4 \times 250 \mathrm{~mm})$; n -hexane-ethyl acetate ( $100: 3 \mathrm{v} / \mathrm{v}$ ), flow rate $1.5 \mathrm{ml} \mathrm{min}{ }^{-1}$ ] to afford the tetramethyl compound (51) $(0.4 \mathrm{mg}, 38 \%)$ as a solid, $\delta_{\mathrm{H}} 0.83,0.86$, and 1.05 (each 3 H , each s, $3 \times \mathrm{Me}$ ) and $1.09(3 \mathrm{H}$, d, $J 7 \mathrm{~Hz}, 2-\mathrm{Me}$ ); $m / z 288\left(M^{+}\right)$(Found: $M^{+}, 288.2452$. Calc. for $\mathrm{C}_{20} \mathrm{H}_{32} \mathrm{O} M, 288.2450$ ), whose n.m.r. spectral data were consistent with those reported. ${ }^{25,26}$
(+)-(3R,4aR,4bS,8aS, 10aS)-3,4,4a,4b,5,6,7,8,8a,9,10,10a-Dodecahydro-4b,8,8-trimethyl-3,10a-ethanophenanthrene (52).To a stirred solution of LDA, prepared from di-isopropylamine ( $5.6 \mathrm{mg}, 0.055 \mathrm{mmol}$ ) and butyl-lithium ( $3.5 \mathrm{mg}, 0.055 \mathrm{mmol}$ ) in dry THF ( 2 ml ), at $-78^{\circ} \mathrm{C}$ under argon was added dropwise a solution of the ketone ( $\mathbf{5 0}$ ) ( $3 \mathrm{mg}, 0.011 \mathrm{mmol}$ ) in dry THF, and the mixture was then stirred for 1 h . After addition of two drops of bromine at $-78^{\circ} \mathrm{C}$, the mixture was further stirred for 0.5 h at this temperature. The mixture was allowed to warm to $0^{\circ} \mathrm{C}$ and was treated with water ( 3 ml ) and extracted with diethyl ether. The extract was washed successively with water and $5 \%$ aqueous sodium thiosulphate, dried, and evaporated to give the bromide ( $3.9 \mathrm{mg}, 100 \%$ ) as an oil, $v_{\text {max. }} 1725 \mathrm{~cm}^{-1}$ (C=O); $m / z$ $353\left(M^{+}\right)$and $355\left(M^{+}+2\right)$.

To a stirred solution of the above bromide ( 3.9 mg ) in ethanol ( 1 ml ) was added sodium borohydride ( 1 mg ). After evaporation of the solvent, the residue was partitioned between saturated aqueous sodium chloride and methylene dichloride. The organic layer was dried and evaporated to give the crude $\alpha$ bromo alcohol, which was used in the next reaction without purification.

A mixture of the above product and zinc ( 12 mg ) in ethanol $(1 \mathrm{ml})$ was refluxed for 1 h , and then the solvent was evaporated off. The residue was taken up into diethyl ether and the solution was filtered through Celite. Evaporation of the filtrate, followed by purification of the residue by h.p.l.c. [LiChrosorb. SI 60 ( $4 \times 250 \mathrm{~mm}$ ); n-hexane, flow rate $\left.1.5 \mathrm{ml} \mathrm{min}^{-1}\right]$ afforded the olefin (52) [ $1.98 \mathrm{mg}, 70 \%$ from (50)] as prisms, m.p. $83-84{ }^{\circ} \mathrm{C}$ (lit. ${ }^{3} 82-83{ }^{\circ} \mathrm{C}$ ); $[x]_{\mathrm{D}}^{24}+16.4^{\circ}\left(c 0.134\right.$ in MeOH ); $\delta_{\mathrm{H}} 0.81$, 0.86 , and 0.95 (each 3 H , each s, $3 \times \mathrm{Me}$ ), $1.00-2.45(19 \mathrm{H}, \mathrm{m}$, $8 \times \mathrm{CH}_{2}$ and $3 \times \mathrm{CH}$ ), and $5.91-6.02(2 \mathrm{H}, \mathrm{m}, 1-$ and $2-\mathrm{H})$.

## Acknowledgments

We thank Mr. K. Kawamura, Miss E. Kurosawa, Miss K. Mushiake, Miss E. Koike, and Miss H. Tanaka of this Institute for microanalysis, spectral measurements, and preparation of the manuscript.

## References

1 A. H. Kapadi, R. R. Sobti, and S. Dev, Tetrahedron Lett., 1965, 2729; A. H. Kapadi, R. Soman, R. R. Sobti, and S. Dev, Indian J. Chem., Sect. B, 1983, 22, 964.
2 R. A. Bell and R. E. Ireland, Tetrahedron Lett., 1963, 269; R. A. Bell, R. E. Ireland, and R. A. Partyka, J. Org. Chem., 1966, 31, 2530.

3 L. H. Zalkow and N. N. Girota, J. Org. Chem., 1964, 29, 1299.
4 R. A. Appleton, P. A. Gunn, and R. McCrindle, Chem. Commun., 1968, 1131.

5 R. M. Coates and E. F. Bertram, J. Org. Chem., 1971, 36, 2625.
6 M. Ihara, M. Toyota, K. Fukumoto, and T. Kametani, Tetrahedron Lett., 1984, 25, 2167; M. Ihara, M. Toyota, M. Abe, Y. Ishida, K. Fukumoto, and T. Kametani, J. Chem. Soc., Perkin Trans. 1, 1986, 1543.

7 Part of this work has been published as preliminary communications; M. Ihara, M. Toyota, K. Fukumoto, and T. Kametani, Tetrahedron Lett., 1984, 25, 3235; 1985, 26, 1537.
8 B. R. Davis and S. J. Johnson, J. Chem. Soc., Perkin Trans. 1, 1979, 2840.

9 W. Nagata and M. Yoshioka, Org. React., 1977, 25, 255.
10 W. Nagata, M. Narisada, and T. Sugasawa, J. Chem. Soc. C, 1967, 648.
11 C. L. Liotta, A. M. Dabdoub, and L. H. Zalkow, Tetrahedron Lett., 1977, 1117.
12 W. Nagata, M. Yoshioka, and S. Hirai, J. Am. Chem. Soc., 1972, 94, 4635.

13 W. Nagata and M. Yoshioka, Org. Synth., 1972, 52, 100.
14 V. Balogh, M. Fétizon, and M. Golfier, J. Org. Chem., 1971, 36, 1339.
15 W. S. Wadsworth, Jr., and W. D. Emmons, Org. Synth., 1965, $45,44$. 16 B. O. Lindgren and T. Nilsson, Acta Chem. Scand., 1973, 27, 888.
17 M. C. Baird, J. T. Mague, J. A. Osborn, and G. Wilkinson, J. Chem. Soc. A, 1967, 1347; K. Ohno and J. Tsuji, J. Am. Chem. Soc., 1968, 90, 99.

18 P. A. Gunn, R. McCrindle, and R. C. Roy, J. Chem. Soc. C, 1971, 1018; J. MacMillan and E. R. H. Walker, J. Chem. Soc. C, 1972, 1274.
19 J. Gutzwiller, P. Buckschacker, and A. Fürst, Synthesis, 1977, 167.
20 F. Sondheimer and D. Elad, J. Am. Chem. Soc., 1957, 79, 5542.
21 M. Numazawa and M. Nagaoka, J. Org. Chem., 1982, 47, 4024.
22 P. A. Grieco, S. Gilman, and M. Nishizawa, J. Org. Chem., 1976, 41, 1485; K. B. Sharpless and M. W. Young, ibid., 1975, 40, 947.
23 R. Pappo, D. S. Allen, Jr., R. V. Lemieux, and W. S. Johnson, J. Org. Chem., 1956, 21, 478.
24 Y. Ito, T. Hirao, and T. Saegusa, J. Org. Chem., 1978, 43, 1011.
25 K. M. Baker, L. H. Briggs, J. G. St. C. Buchanan, R. C. Cambie, B. R. Davis, R. C. Hayward, G. A. S. Long, and P. S. Rutledge, J. Chem. Soc., Perkin Trans. 1, 1972, 190.
26 J. MacMillan and E. R. H. Walker, J. Chem. Soc., Perkin Trans. 1, 1972, 1274.
27 L. F. Fieser and R. Ettorre, J. Am. Chem. Soc., 1953, 75, 1700; M. R. Roberts and R. H. Schlessinger, ibid., 1981, 103, 724.

Received 3rd January 1986; Paper 6/012

